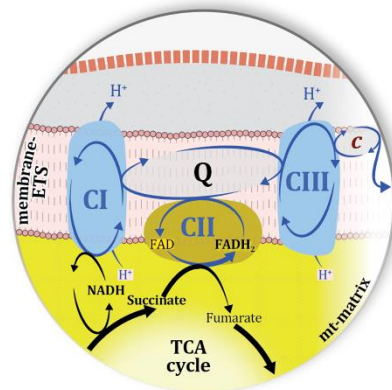


Theoretical Communication

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 succinate dehydrogenase, SDH
 tricarboxylic acid cycle, TCA

Complex II ambiguities – FADH₂ in the electron transfer system

 Erich Gnaiger

Oroboros Instruments, Innsbruck, Austria.

Correspondence: erich.gnaiger@orooboros.at

Summary

The current narrative that the reduced coenzymes NADH and FADH₂ feed electrons from the tricarboxylic acid cycle into the mitochondrial electron transfer system creates ambiguities around respiratory Complex II (CII). The succinate dehydrogenase subunit SDHA of CII oxidizes succinate and reduces covalently bound FAD to FADH₂ in the canonical forward tricarboxylic acid cycle. However, several graphical representations of the membrane-bound electron transfer system (ETS) depict FADH₂ in the mitochondrial matrix to be oxidized by CII. This leads to the false conclusion that FADH₂ feeds electrons into the ETS through CII, including FADH₂ from the tricarboxylic acid cycle and the β-oxidation cycle in fatty acid oxidation. In reality, FAD and succinate are the *substrates* of SDHA at the ETS-entry into CII. The reduced flavin groups FADH₂ and FMNH₂ are *products* downstream within CII and CI, respectively. Further electron transfer converges at the coenzyme Q-junction. Similarly, electron transferring flavoprotein and mitochondrial glycerophosphate dehydrogenase feed electrons into the Q-junction but not through CII. The ambiguities surrounding Complex II in the literature and educational tools call for quality control, to secure scientific standards in current communications on bioenergetics and ultimately support adequate clinical applications.

1. Introduction

The tricarboxylic acid (TCA) cycle – the citric acid cycle or Krebs cycle – sparked a renaissance of interest in cellular and mitochondrial bioenergetics (Gnaiger et al 2020; Bénit et al 2022; Arnold, Finley 2023). TCA cycle metabolites are oxidized while reducing NAD⁺ to NADH in the forward cycle, or are transported into the cytosol (Murphy, O'Neill 2018). Respiratory Complex II (CII, succinate dehydrogenase SDH; succinate-ubiquinone oxidoreductase; EC 1.3.5.1) has a unique position in both the TCA cycle and the mitochondrial membrane-bound electron transfer system (membrane-ETS). All genes for

CII are nuclear encoded, with exceptions in red algae and land plants (Huang et al 2019; Moosavi et al 2019). Succinate:quinone oxidoreductases (SQRs, succinate dehydrogenases SDH) favour oxidation of succinate and reduction of quinone in the canonical forward direction of the TCA cycle and electron transfer into the Q-junction (Cecchini 2003). Operating in the reverse direction, quinol:fumarate reductases (QFRs, fumarate reductases, FRD) reduce fumarate and oxidize quinol (Iverson 2013; Maklashina et al 2022). The reversed TCA cycle has gained interest in studies ranging from metabolism in anaerobic animals (Hochachka, Somero 2002), thermodynamic efficiency of anaerobic and aerobic ATP production (Gnaiger 1993), reversed electron transfer and production of reactive oxygen species (Tretter et al 2016; Robb et al 2018; Spinelli et al 2021), hypoxia and ischemia-reperfusion injury (Couchani et al 2014), to evolution of metabolic pathways (Lane 2022). In cancer tissue CII plays a key role in metabolic remodeling (DeBerardinis, Chandel 2016; Schöpf et al 2020).

Two-electron transfer $2\{e^{-}\}$ from succinate to the oxidized flavin adenoside dinucleotide FAD is redox-coupled to the transfer of two hydrogen ions $2\{H^{+}\}$ with formation of FADH₂. This H⁺-linked electron transfer (Hsu et al 2022) through CII is not coupled to H⁺ translocation across the mitochondrial inner membrane (mtIM). Hence, CII is not a H⁺ pump in contrast to the respiratory Complexes CI, CIII and CIV through which electron transfer drives and maintains the protonmotive force. The coenzyme NAD⁺ is reduced to NADH+H⁺ during the oxidation of pyruvate and through redox reactions catalyzed by TCA cycle enzymes including isocitrate dehydrogenase, oxoglutarate (α -ketoglutarate) dehydrogenase, and malate dehydrogenase. In turn, coenzyme FAD is reduced to FADH₂ during oxidation of succinate by succinate dehydrogenase (CII). Confusion emerges, however, when NADH and FADH₂ are considered as the reduced compounds feeding electrons from the TCA cycle into the ‘respiratory chain’ – rather than NADH and succinate (Gnaiger 2020). This ‘Complex II ambiguity’ has deeply penetrated the scientific literature on bioenergetics without sufficient quality control. Therefore, a critical literature survey is needed to draw attention to widespread ambiguities, particularly in graphical representations of the mitochondrial electron transfer system, to ensure scientific standards in communications on bioenergetics.

2. Experimental evidence

Complex II is a flavoprotein with a covalently bound flavin adenine dinucleotide as documented in early reports (Kearney 1960) and summarized in classical textbooks (Lehninger 1970; Tzagoloff 1982). Microscopic detail on the structure and function of CII has expanded our knowledge on the mechanism of enzyme assembly (Maklashina et al 2022), enzyme structure (Vercellino, Sazanov 2022), kinetic regulation of CII activity (Mills et al 2018; Fink et al 2022), and associated pathologies (Bénil et al 2022).

The reversible oxidoreduction of succinate and fumarate is catalyzed in the soluble domain of CII extending from the mtIM into the mt-matrix. Succinate donates electrons – i.e. two hydrogen ions and two electrons ($2\{H^{+}+e^{-}\}$) – to the cofactor FAD which is tightly bound to the subunit SDHA. SDHA contains the catalytically active dicarboxylate binding site where succinate is oxidized to fumarate. The oxidized yellow (450 nm) form FAD functions as hydrogen acceptor from succinate to the reduced product FADH₂ while fumarate is formed as the oxidized product in the TCA cycle. Like in most flavin-linked dehydrogenases, the flavin nucleotide remains tightly bound to the enzyme during the

catalytic cycle. FADH₂ relays electrons further through a series of iron-sulfur redox centers in SDHB to ubiquinone in the membrane domain harboring SDHC and SDHD (Moosavi et al 2019) (Figure 1a).

The reduced flavin groups FADH₂ of flavin adenine dinucleotide and FMNH₂ of flavin mononucleotide are at functionally comparable levels in the electron transfer in CII and CI, respectively, to the Q-junction (Figure 1b). FMN in CI is reduced by NADH forming (reduced) FMNH₂ and (oxidized) NAD⁺. FADH₂ and FMNH₂ are reoxidized downstream in CII and CI, respectively, by final electron transfer to coenzyme Q (Figure 1b).

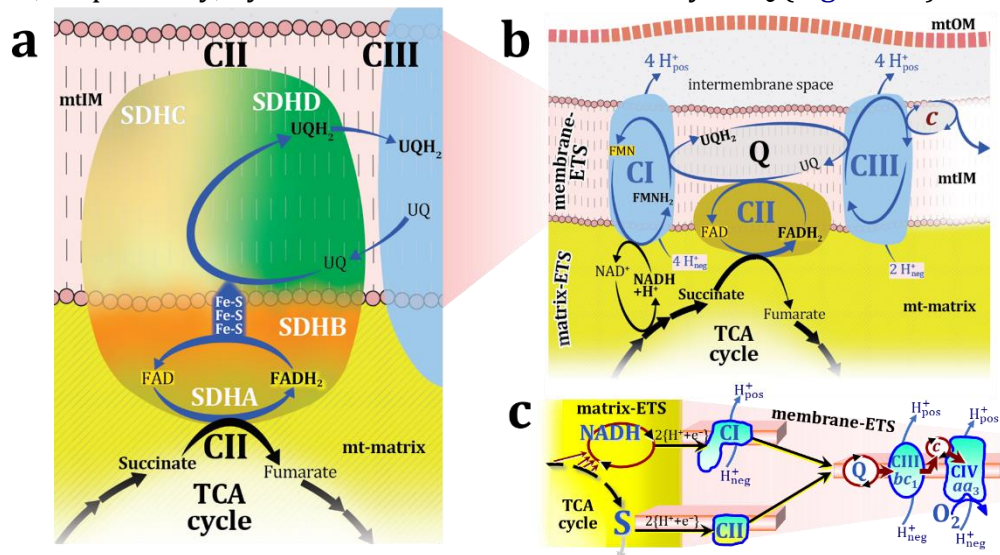


Figure 1. Complex II bridges electron transfer from the TCA cycle to the mitochondrial inner membrane. Graphical representations of the electron transfer system ETS with successive emphasis on pathway architecture and concomitant loss of detail. CII is integrated in the TCA cycle (matrix-ETS) and the membrane-bound electron transfer system (membrane-ETS of the mt-inner membrane mtIM). Joint pairs of half-circular arrows indicate electron transfer $2\{H^{+}+e^{-}\}$, distinguished from vectorial hydrogen ion transport across the mtIM ($H^{+}_{neg} \rightarrow H^{+}_{pos}$) from the negatively to the positively charged compartment. **(a)** In the soluble domain of CII, the flavoprotein SDHA catalyzes the oxidation succinate \rightarrow fumarate $+2\{H^{+}+e^{-}\}$ and reduction FAD $+2\{H^{+}+e^{-}\} \rightarrow$ FADH₂. The iron-sulfur protein SDHB transfers electrons through Fe-S clusters to the mtIM domain where ubiquinone UQ is reduced with $2\{H^{+}+e^{-}\}$ to ubiquinol UQH₂ in SDHC and SDHD. **(b)** NADH and succinate are substrates of $2\{H^{+}+e^{-}\}$ transfer to CI and CII, respectively, with FMNH₂ and FADH₂ as the corresponding products. NADH+H⁺ and NAD⁺ cycle between matrix-dehydrogenases and CI, whereas FAD and FADH₂ cycle within the enzyme CII. Succinate and fumarate indicate the chemical entities irrespective of ionization, whereas the charges are shown in NADH, NAD⁺, and H⁺. **(c)** Electron flow catalyzed by dehydrogenases localized in the mt-matrix converges at the N-junction, reducing NAD⁺ to NADH. Electron flow from NADH and succinate S to molecular oxygen, $2\{H^{+}+e^{-}\}+0.5 O_2 \rightarrow H_2O$, converges through CI and CII at the Q-junction. CIII passes electrons to cytochrome c and in CIV to O₂.

The branches of electron transfer from NADH and succinate converge through CI and CII at the Q-junction. The convergent architecture of the electron transfer system (ETS; in contrast to a linear electron transfer chain) is emphasized in Figure 1c (Hatefi 1962; Gnaiger 2020). Comparable to CII, several additional respiratory Complexes are

localized in the mtIM which catalyze electron transfer converging at the Q-junction, including electron transferring flavoprotein (ETF) in fatty acid oxidation, glycerophosphate dehydrogenase (GpDH), sulfide-ubiquinone oxidoreductase, choline dehydrogenase, dihydro-ototat dehydrogenase, and proline dehydrogenase (Gnaiger 2020; Bénit et al 2022; Pallag et al 2022).

3. Results and discussion

3.1. The source and consequence of Complex II ambiguities

'No representation is ever perfectly expressive, for if it were it would not be a representation but the thing itself' (Grosholz 2007).

Ambiguities emerge if the representation of a concept is vague to an extent that allows for equivocal interpretations. As a consequence, even a basically clear concept (Figure 1) may be communicated as a divergence from an established truth. The comparison between NADH linked to CI and FADH₂ (instead of succinate) linked to CII leads us astray, as illustrated by the following quotes from Cooper (2000) (Figure 2).

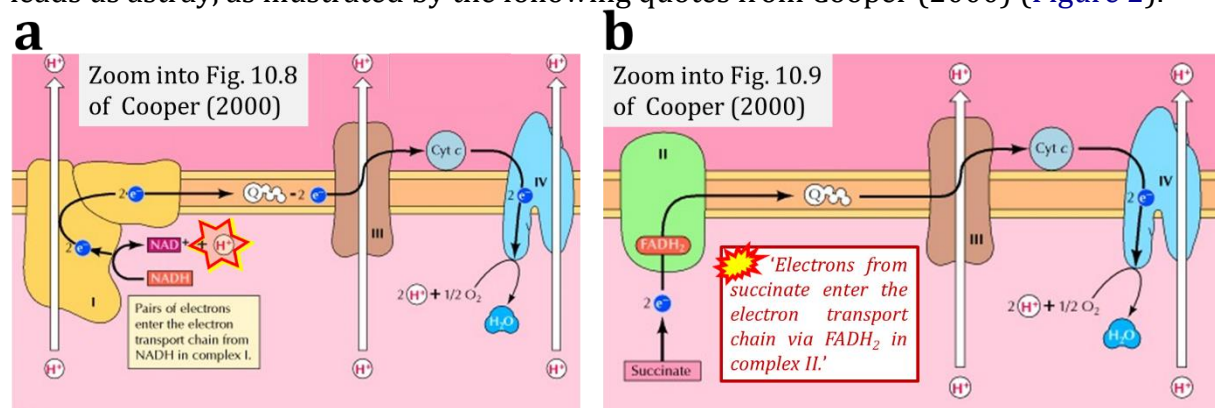


Figure 2. Electron flow into Complexes (a) CI and (b) CII. Zoom into figures of Cooper (2000). **(a)** The marked H⁺ is consumed in H⁺-linked electron transfer instead of being produced. **(b)** Marked quote inserted from the legend to Figure 10.9.

(1) *'Electrons from NADH enter the electron transport chain in complex I, .. A distinct protein complex (complex II), which consists of four polypeptides, receives electrons from the citric acid cycle intermediate, succinate (Figure 10.9). These electrons are transferred to FADH₂, rather than to NADH, and then to coenzyme Q.'*

(2) *'In contrast to the transfer of electrons from NADH to coenzyme Q at complex I, the transfer of electrons from FADH₂ to coenzyme Q is not associated with a significant decrease in free energy and, therefore, is not coupled to ATP synthesis.'* Note that CI is in the path of electron transfer from NADH to coenzyme Q. In contrast, electron transfer from FADH₂ to coenzyme Q is downstream of succinate oxidation by CII. Thus even a large Gibbs force ('decrease in free energy') in FADH₂→Q would fail to drive the coupled process of proton translocation through CII. The total Gibbs force (Gnaiger 2020) in S→FADH₂→Q must be accounted for. (In parentheses: None of these steps are directly coupled to ATP synthesis. Redox-driven proton translocation must be distinguished from phosphorylation of ADP driven by the protonmotive force).

(3) CII receives electrons from succinate, yet it is suggested that *'electrons from succinate enter the electron transport chain via FADH₂ in complex II.'* The ambiguity is

caused by a lack of unequivocal definition of the electron transfer system ('*electron transport chain*'; [Supplement 1](#)). Two contrasting definitions are implied of the '*electron transport chain*' or ETS. (a) CII is part of the ETS. Hence electrons enter the ETS from succinate but not from FADH₂ – from the matrix-ETS to the membrane-ETS ([Figure 1b,c](#)). (b) If electrons enter the '*electron transport chain* via FADH₂ in complex II', then subunit SDHA would be upstream and hence not part of the ETS (to which conclusion obviously nobody would agree). There remains the ambiguity of electron entry into CII from succinate ([Figure 1](#)) or from FADH₂ as the product of succinate dehydrogenase in the TCA cycle ([Figure 3](#)).

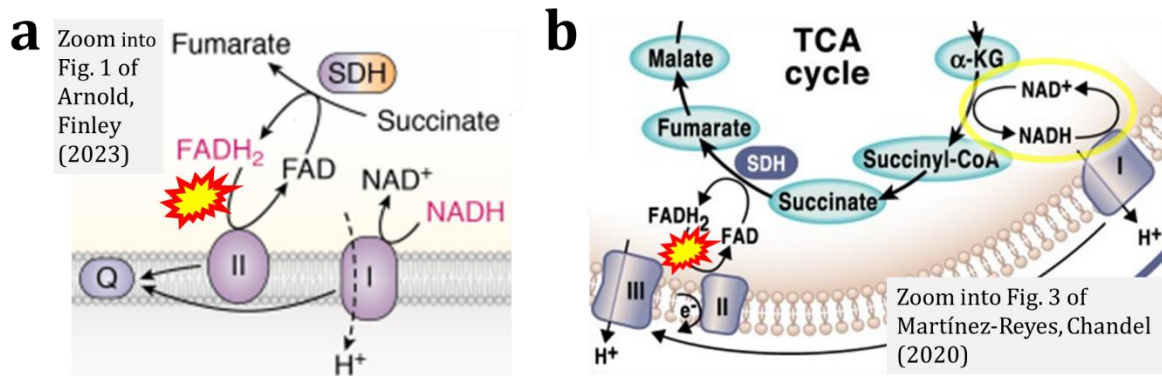


Figure 3. FADH₂ depicted as product and substrate of Complex II. Zoom into figures by (a) Arnold, Finley (2023) and (b) Martínez-Reyes, Chandel (2020). NADH and NAD⁺ cycle between different types of enzymes (yellow circle), in contrast to the FADH₂-FAD cycle located within the same enzyme Complex (SDH and CII are synonyms).

3.2. The FADH₂ - FAD confusion in the succinate-pathway

The narrative that the reduced coenzymes NADH and FADH₂ feed electrons from the TCA cycle into the mitochondrial electron transfer system causes confusion. As a consequence, FADH₂ appears in several publications erroneously as the substrate of CII in the ETS linked to succinate oxidation. This error is widely propagated in 98 publications found from 2001 to 2023 ([Supplements 2 to 6](#)) and educational websites ([Supplement 7](#)). Clarification is required (Gnaiger 2020; page 48). The following examples illustrate the transition from ambiguity to misunderstanding.

(1) Ambiguities appear in graphical representations, where FADH₂ is the product and substrate of CII (synonymous with SDH) in the same figure ([Figure 3](#); [Suppl Figure S2](#)).

(2) Correct representation or ambiguity evolved to misconception in graphical representations ([Figure 4](#)).

(3) Graphical errors on electron entry from FADH₂ into CII show up without comment in or context to the text ([Figure 5](#); [Suppl Figures S3](#)). Instead of NADH+H⁺→NAD⁺ there appears NADH→NAD⁺+H⁺ (or +2H⁺) and by analogy FADH→FAD +2H⁺ ([Figure 5d](#); [Suppl Figure S4](#)). The analogy NADH→NAD⁺ is taken further to include a charge for FAD or even writing FADH⁺ as the product ([Figure 6](#); [Suppl Figure S5](#)).

(4) Error propagation from graphical representation ([Figure 3a](#)) to misunderstanding in the text: '*SDH reduces FAD to FADH₂, which donates its electrons to complex II*'; '*each complete turn of the TCA cycle generates three NADH and one FADH₂ molecules, which donate their electrons to complex I and complex II, respectively*'; '*complex I and complex II oxidize NADH and FADH₂, respectively*' (Arnold, Finley 2023).

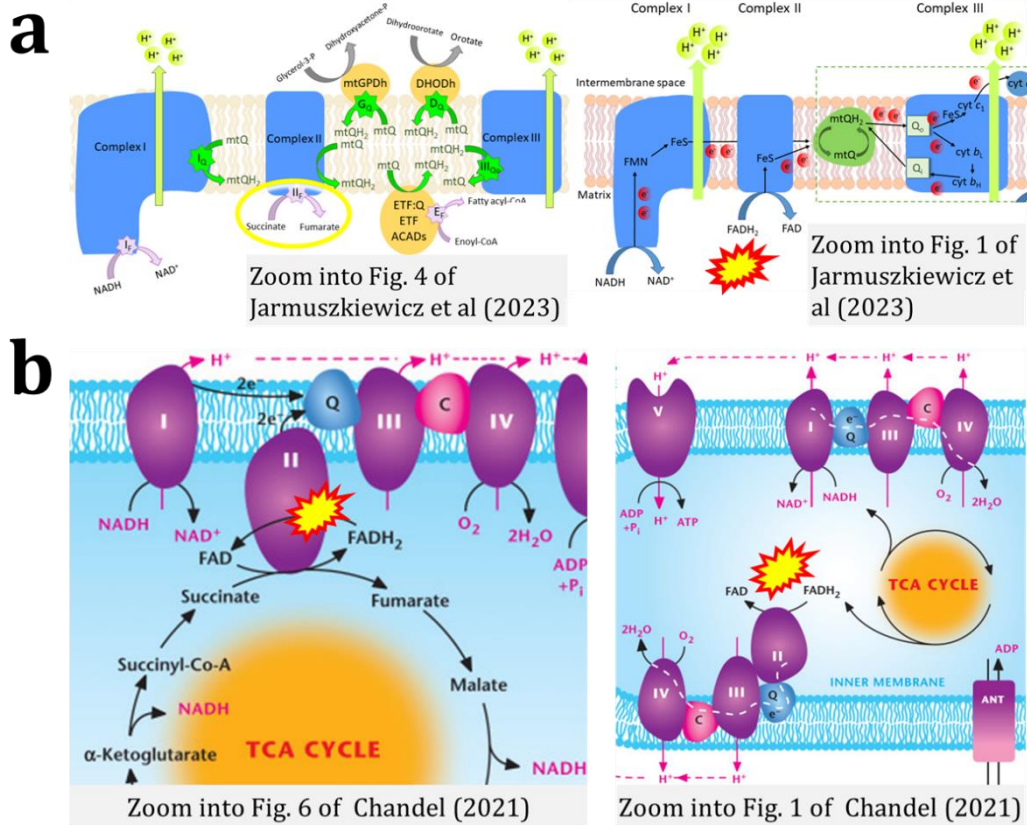


Figure 4. Evolving disarrangement in graphical representations of FADH₂ as a substrate of CII. (a) Succinate or FADH₂ as substrates of CII (Jarmuskiewicz et al 2023). (b) From ambiguity to misconception (Chandel 2021).

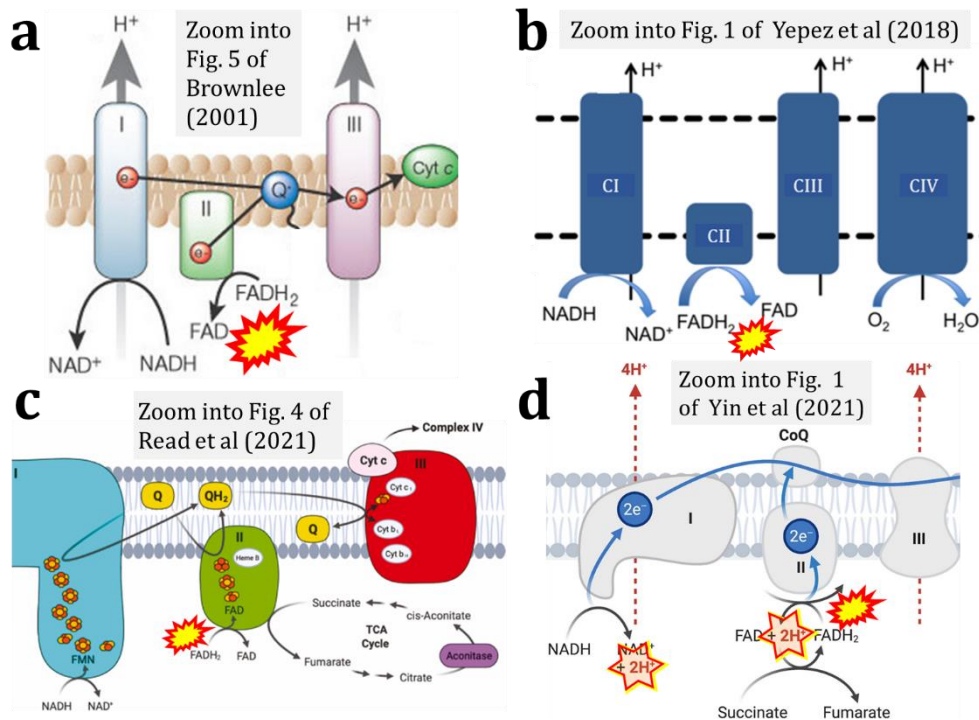


Figure 5. FADH₂ shown as substrate of CII. Zoom into figures from (a) Brownlee (2001); (b) Yépez et al (2018); (c) Read et al (2021) showing FAD as product in CII and the mt-matrix; (d) Yin et al (2021) with unjustified indication of 2H⁺ formation in the mt-matrix.

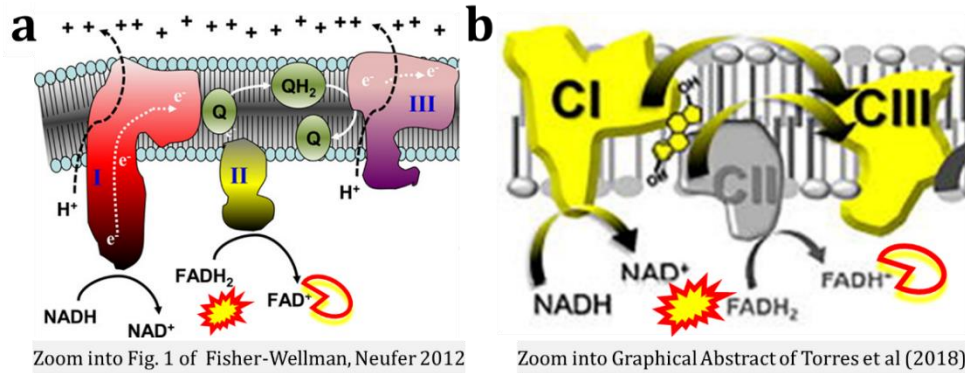


Figure 6. FADH₂ is shown as the substrate of CII. These graphical representations take the NADH→NAD⁺ analogy to the level of depicting FAD as (a) FAD⁺ (Fisher-Wellman, Neuffer 2012) or (b) FADH⁺ (Torres et al 2018).

In summary, downstream of the dehydrogenases of the TCA cycle, NADH is oxidized by CI. Two-electron oxidation of succinate is redox-linked to reduction of FAD to FADH₂. In terms of electron entry into CII many publications show it in the wrong direction, i.e. FADH₂ as electron donor from the TCA cycle to CII (Figures 3 to 6; Suppl Figures S2 to S6).

3.3. Oxidation of FADH₂ to FAD and 2{H⁺+e⁻} transfer

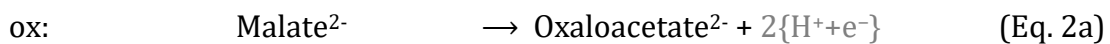
Electron transfer from succinate in the TCA cycle to coenzyme FAD can be written as a redox reaction, where oxidation (ox) of succinate yields two hydrogen ions and two electrons 2{H⁺+e⁻} which are donated in the reduction (red) of FAD to FADH₂,



which yields the net redox reaction equation



Commonly the charges of succinate, fumarate (Eq. 1), and other metabolites are not shown explicitly in graphical representations of metabolic pathways, but NAD⁺ is clearly distinguished from FAD (Figure 1b). Taking oxidation of malate by malate dehydrogenase for comparison,



In brief, oxidation of NADH and FADH₂ may be represented as



H⁺ in Eq. 3a is frequently omitted to simplify graphical representations, and a pair of rounded arrows – one external touching the enzyme and one internal within the enzyme – indicates H⁺-linked electron transfer in terms of 2{H⁺+e⁻} (Figures 1a, 1b, 3a, and 4). However, caution is warranted to distinguish (1) H⁺ in chemical acid/base reactions, such as the hydrogencarbonate equilibrium H₂CO₃ ↔ HCO₃⁻ + H⁺, (2) chemical H⁺-linked electron *transfer* (Hsu et al 2022) indicated as 2{H⁺+e⁻} in redox reactions (Eq. 1 and 2), and (3) coupled vectorial *transport* or translocation of H⁺ across the mtIM (H⁺_{neg} → H⁺_{pos}; Figures 1b and c; Supplement 1). The equilibrium in Eq. 3a depends on pH, whereas Eq.

3b is independent of pH. The fundamental difference between H^+ and $2\{H^++e^-\}$ in Eq. 3a is lost in representations such as Figure 5d.

Disturbingly, oxidation of $FADH_2$ is shown in meaningless patterns in various figures, occasionally corresponding with analogous representations of oxidation of $NADH$ (Figure 6; Table 1).

Table 1. Misconceptions in graphical representations of electron entry into CII.

Analogy with NADH	Suppl Figure	$FADH_2$	Suppl Figure
$NADH + H^+ \rightarrow NAD^+$			
$NADH \rightarrow NAD^+ + H^+$	S3d,r,λ,π,φ	$FADH_2 \rightarrow FAD$	S2, S3
$NADH \rightarrow NAD^+ + H^+$	S4a,e,g	$FADH_2 \rightarrow FAD + 2H^+$	S4a-i
$NADH \rightarrow NAD^+ + 2H^+$	S4c,f,h,i		
		$FADH_2 \rightarrow FAD^+$	S5a-g
$NADH + H^+ \rightarrow NADH$	S6a	$FADH_2 \rightarrow FADH$	S6a-d
		$FADH_2 \rightarrow FADH^+$	S6e
		$FADH \rightarrow$	S6f
$NADH \rightarrow NAD + H^+$	S4b	$FADH \rightarrow FAD^+$	S6g
$NADH \rightarrow NAD^+ + H^+$	S6h	$FADH \rightarrow FAD^+ + H^+$	S6h
$NADH \rightarrow NAD^+ + H^+$	S6i	$FADH \rightarrow FAD^+ + 2H^+$	S6i

The erroneous presentation of electron transfer from $FADH_2$ to CII has a logical consequence. β -oxidation generates $FADH_2$ (Figure 7). If $FADH_2$ would donate electrons to CII, then CII can be seen as an enzyme involved downstream of $FADH_2$ in FAO. This topic requires clarification.

3.4. Complex II and fatty acid oxidation

Electron transferring flavoprotein CETF and CI are the respiratory Complexes involved in convergent electron entry into the Q-junction during FAO (Figure 7).

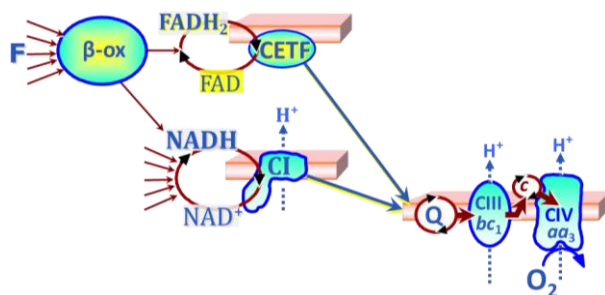


Figure 7. Fatty acid oxidation through the β -oxidation cycle (β -ox), electron transferring flavoprotein (CETF), and Complex I (CI) with convergent electron transfer into the Q-junction. Modified after Gnaiger (2020).

In the β -oxidation cycle of FAO, acetyl-CoA and the reducing equivalents $FADH_2$ and $NADH$ are formed in reactions catalyzed by acyl-CoA dehydrogenases and hydroxyacyl-CoA dehydrogenases, respectively, in the mitochondrial matrix (Houten et al 2016). When $FADH_2$ is erroneously shown as a substrate of CII, a dubious role of CII in FAO is suggested as a consequence (Figure 8; Supplement 8). Lemmi et al (1990) noted: 'mitochondrial Complex II also participates in the oxidation of fatty acids'. This holds for the oxidation of acetyl-Co in the TCA cycle, forming $NADH$ and succinate with downstream electron flow through CI and CII, respectively, into the Q-junction (Figure 1). In contrast, electron transfer from $FADH_2$ formed during β -oxidation proceeds through electron transferring flavoprotein CETF and entry into the Q-junction independent of CII (Figure 7). Fatty acylCoA dehydrogenase reduces FAD to $FADH_2$ in the mitochondrial matrix. The

FADH₂ of the of the fatty acyl-CoA dehydrogenase is reoxidized by the FAD-containing electron transfer flavoprotein Complex CETF. Thus FADH₂ can be seen as an internal substrate of CETF, comparable to NADH as a substrate of CI, succinate as a substrate of CII and glycerophosphate as a substrate of CGpDH.

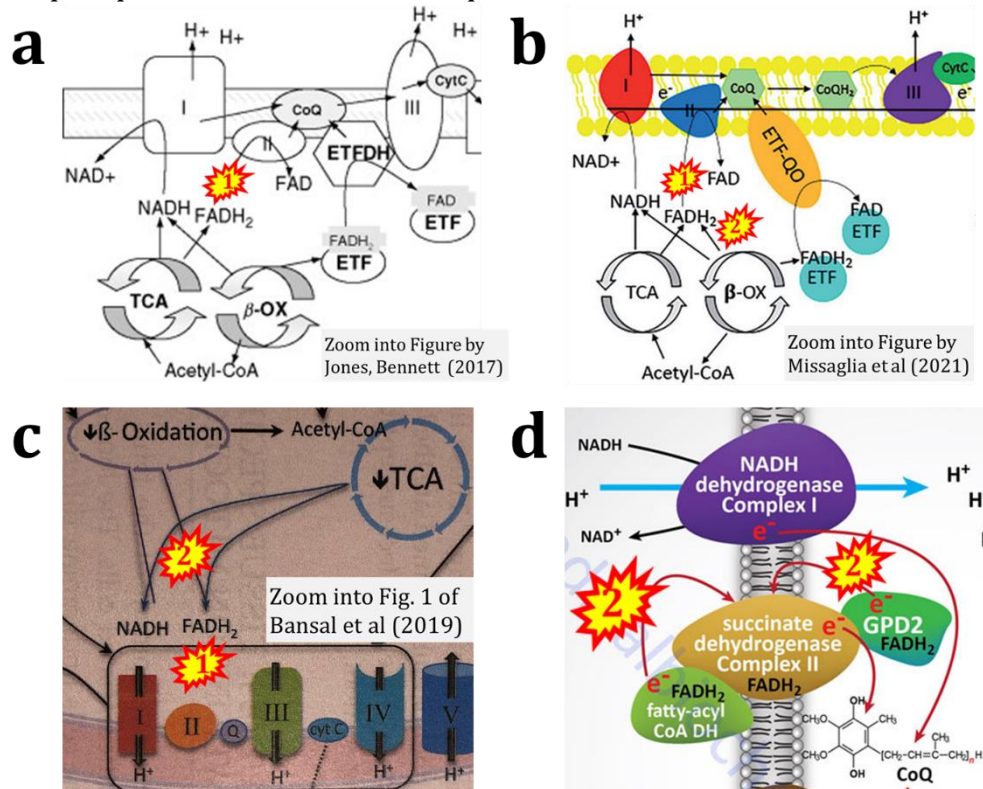


Figure 8. When FADH₂ is erroneously shown as a substrate of CII (1), a role of CII in oxidation of FADH₂ from fatty acid oxidation is suggested as a consequence (2). Zoom into figures from (a) Jones, Bennett (2017); (b) Missaglia et al (2021); (c) Bansal et al (2019); (d) <https://themedicalbiochemistrypage.org/oxidative-phosphorylation-related-mitochondrial-functions/> (accessed 2023-03-21).

4. Conclusions

There is currently ambiguity surrounding the precise role of Complex II in fatty acid oxidation. While Complex II is not essential for fatty acid oxidation, it plays a regulatory role by sensing changes in metabolic demand and activating the TCA cycle for oxidation of acetyl-Co depending on the metabolic conditions. This regulatory function may be particularly important during periods of low oxygen availability or high energy demand. The integration of FAO with the membrane-bound ETS (Wang et al 2019) has significant implications for understanding and treating disorders related to β -oxidation and oxidative phosphorylation. Using precisely defined terminology can prevent misunderstandings (Gnaiger et al 2020; footnotes in Supplement 1). Do erroneous diagrams – from ambiguous electron transfer (Suppl Figures S2 to S8) to presentation of CII as a H⁺ pump (Suppl Figure S9) – cast some doubts on the quality of the publication? Whether using iconic or symbolic elements in graphical representations, incorporating complementary text not only enhances the communication of intended meaning but diagrams will be improved in the process. When peer review provides insufficient help for corrections, post-peer review by editors and critical readers is required for revisions

of articles which may be re-published as living communications (Gnaiger 2021). Clarification instead of perpetuation of Complex II ambiguities leads to a better representation of fundamental concepts of bioenergetics and helps to maintain the high scientific standards required for translating knowledge on metabolism into clinical solutions for mitochondrial diseases.

Abbreviations

2{H ⁺ +e ⁻ }	redox equivalents in electron transfer	NADH ₂	reduced nicotinamide adenine dinucleotide
CI	Complex I	NAD ⁺	oxidized nicotinamide adenine dinucleotide
CII	Complex II	Q	ETS-reactive coenzyme Q, oxidation state is not implied
CETF	electron transferring flavoprotein	QFR	mena-quinol-fumarate oxidoreductase
FADH ₂	reduced flavin adenoside dinucleotide	SQR	succinate-ubiquinone oxidoreductase
FAD	oxidized flavin adenoside dinucleotide	SDH, SDHABCD	succinate dehydrogenase, CII
FAO	fatty acid oxidation	TCA cycle	tricarboxylic acid cycle
FMNH ₂	reduced flavin mononucleotide		
mt-matrix	mitochondrial matrix		
mtIM	mitochondrial inner membrane		

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Supplement 1

Footnotes on terminology

Electron transfer system ETS: The *convergent* architecture of the electron transfer *system* is emphasized in contrast to *linear* electron transfer *chains* ETCs within segments of the ETS.

Electron transfer: A distinction is necessary between electron *transfer* in redox reactions and electron *transport* (translocation) in the diffusion of charged ionic species within or between cellular compartments. The symbol $2\{H^+e^-\}$ is introduced to indicate H⁺-linked electron transfer of two hydrogen ions and two electrons in a redox reaction.

H⁺-linked electron transfer: The term H⁺-coupled electron transfer (Hsu et al 2022) is replaced by H⁺-*linked* electron transfer, to avoid confusion with *coupled* H⁺ translocation.

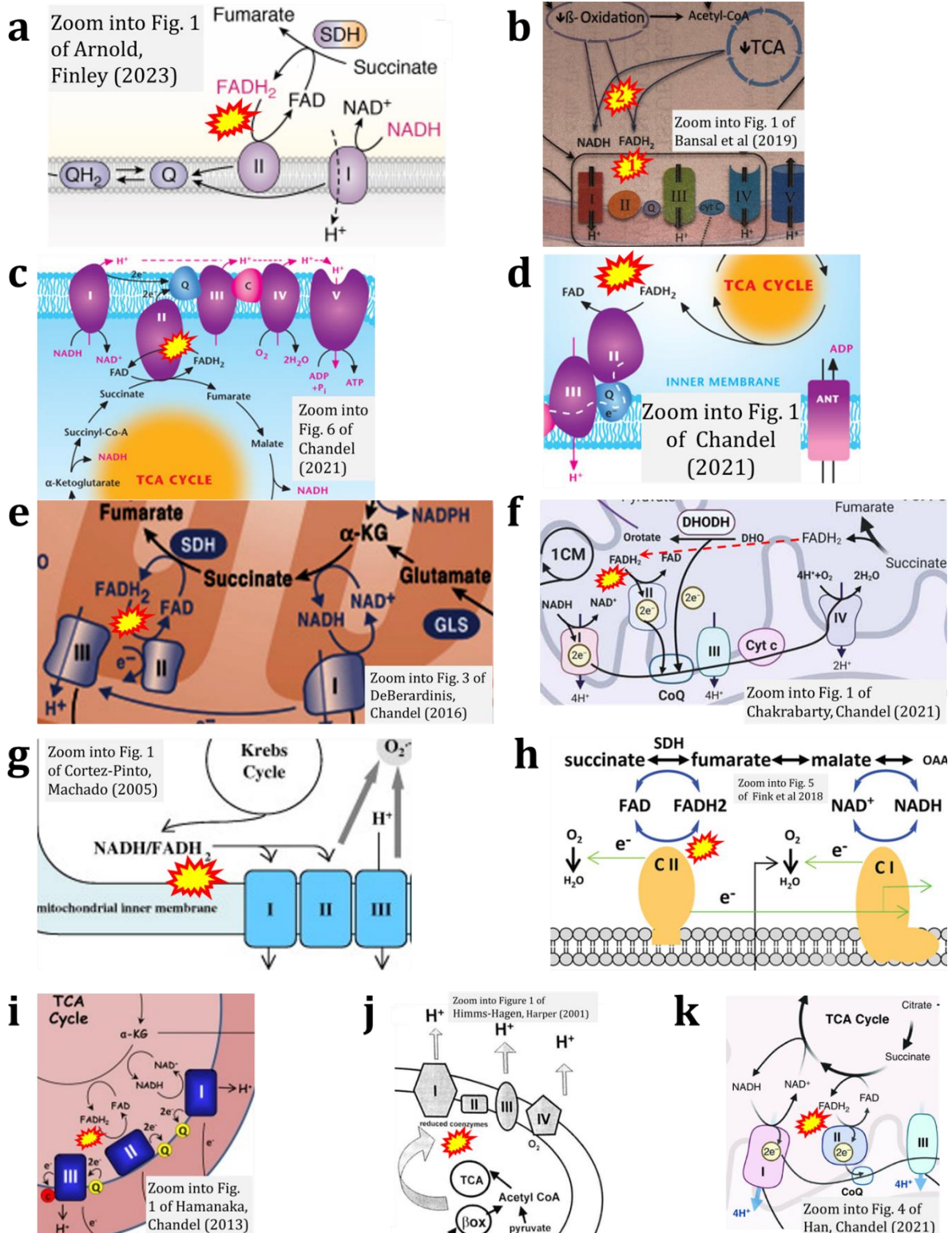
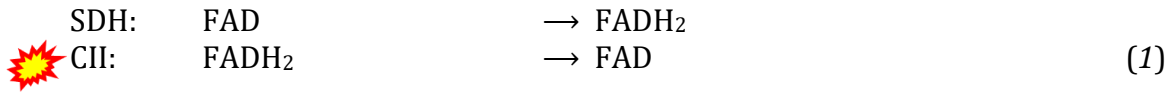
Matrix-ETS: Electron transfer and corresponding OXPHOS capacities are classically studied in mitochondrial preparations as oxygen consumption supported by various fuel substrates undergoing partial oxidation in the mt-matrix, such as pyruvate, malate, succinate, and others. Therefore, the *matrix* component of ETS (matrix-ETS) is distinguished from the ETS *bound to the mt-inner membrane* (membrane-ETS; Gnaiger et al 2020).

Membrane-ETS: Electron transfer is frequently considered as the segment of redox reactions linked to the mtIM. However, the *membrane*-ETS is only part of the total ETS, which includes the upstream *matrix*-ETS.

$2\{H^+e^-\}$: The symbol $[2 H]$ is frequently used to indicate redox equivalents in the transfer from hydrogen donors to hydrogen acceptors. However, $2[H]$ does not explicitly express that it applies to both *electron* and *hydrogen ion* transfer. Brackets are avoided to exclude the confusion with their frequent application to indicate amount-of-substance concentrations.

Supplement 2

FAD a substrate of SDH and FADH₂ a substrate of CII (Figure S2)



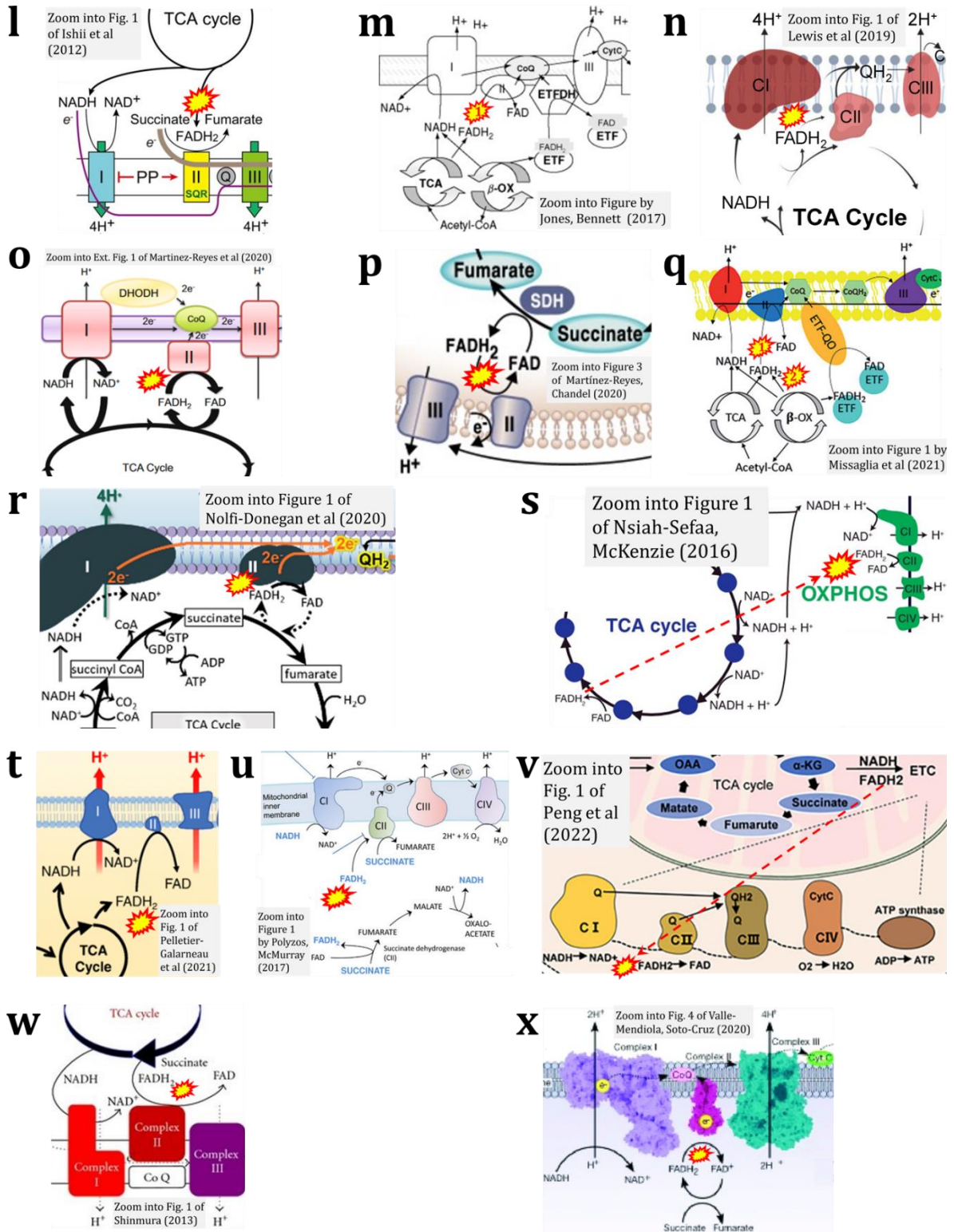


Figure S2. Complex II ambiguities in graphical representations on FADH₂ as a substrate of Complex II in the canonical forward electron transfer. The TCA cycle reduces FAD to FADH₂ - in several cases shown to be catalyzed by SDH. Then FADH₂ is erroneously shown to feed electrons into CII. Alphabetical sequence of publications from 2001 to 2023. See References for Figure S2.

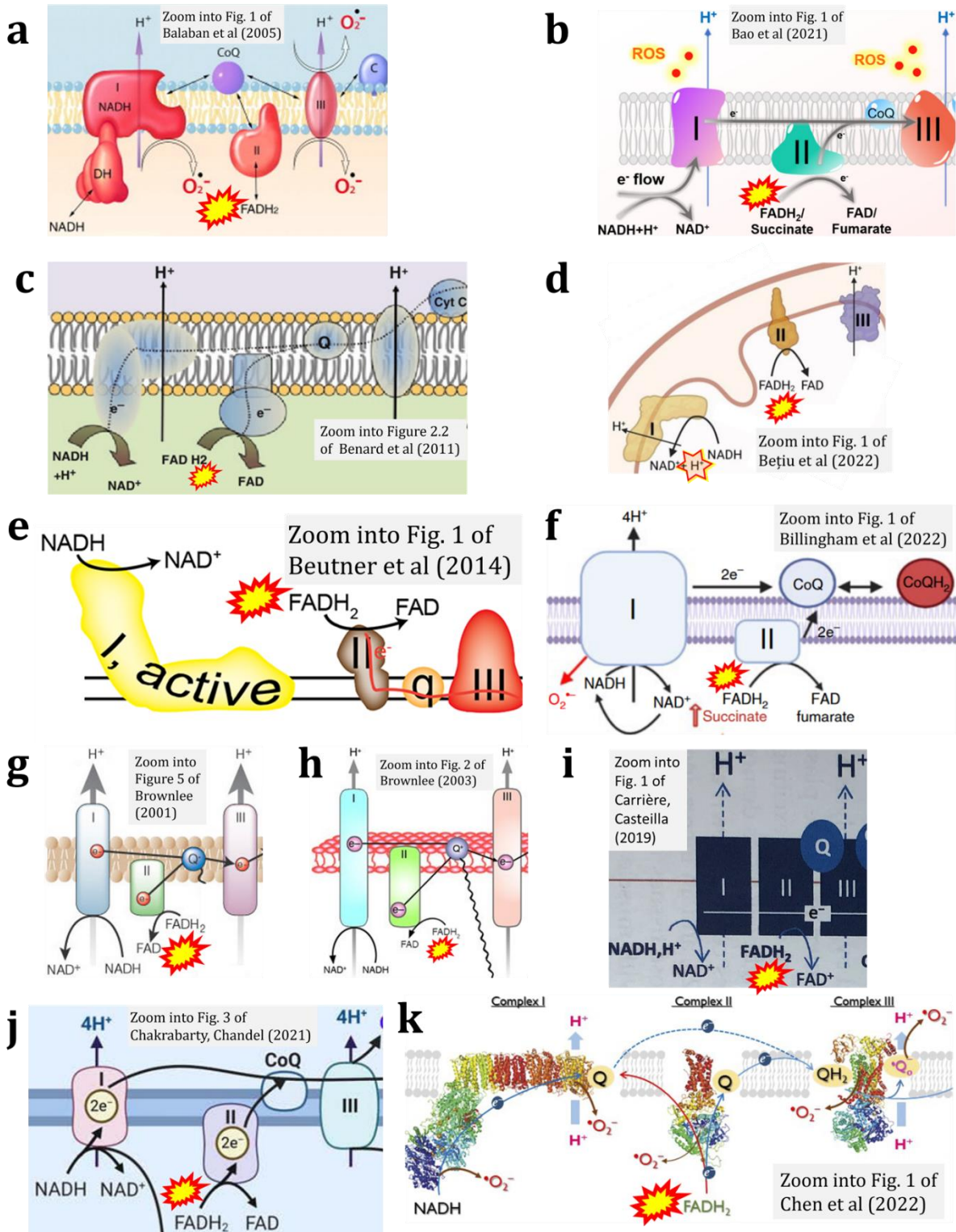
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Supplement 3

FADH₂ as substrate of CII (Figure S3)



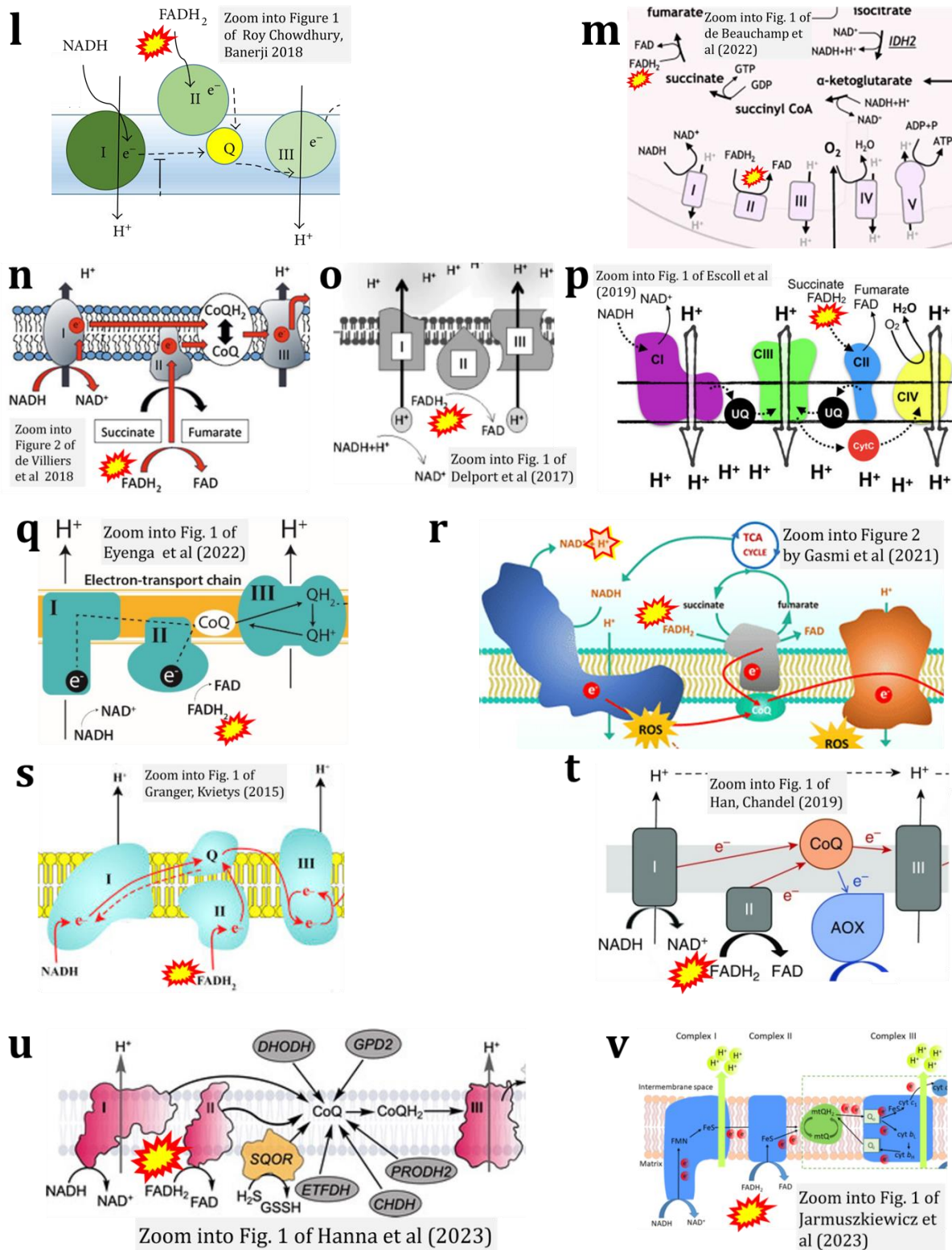


Figure S3. Continued

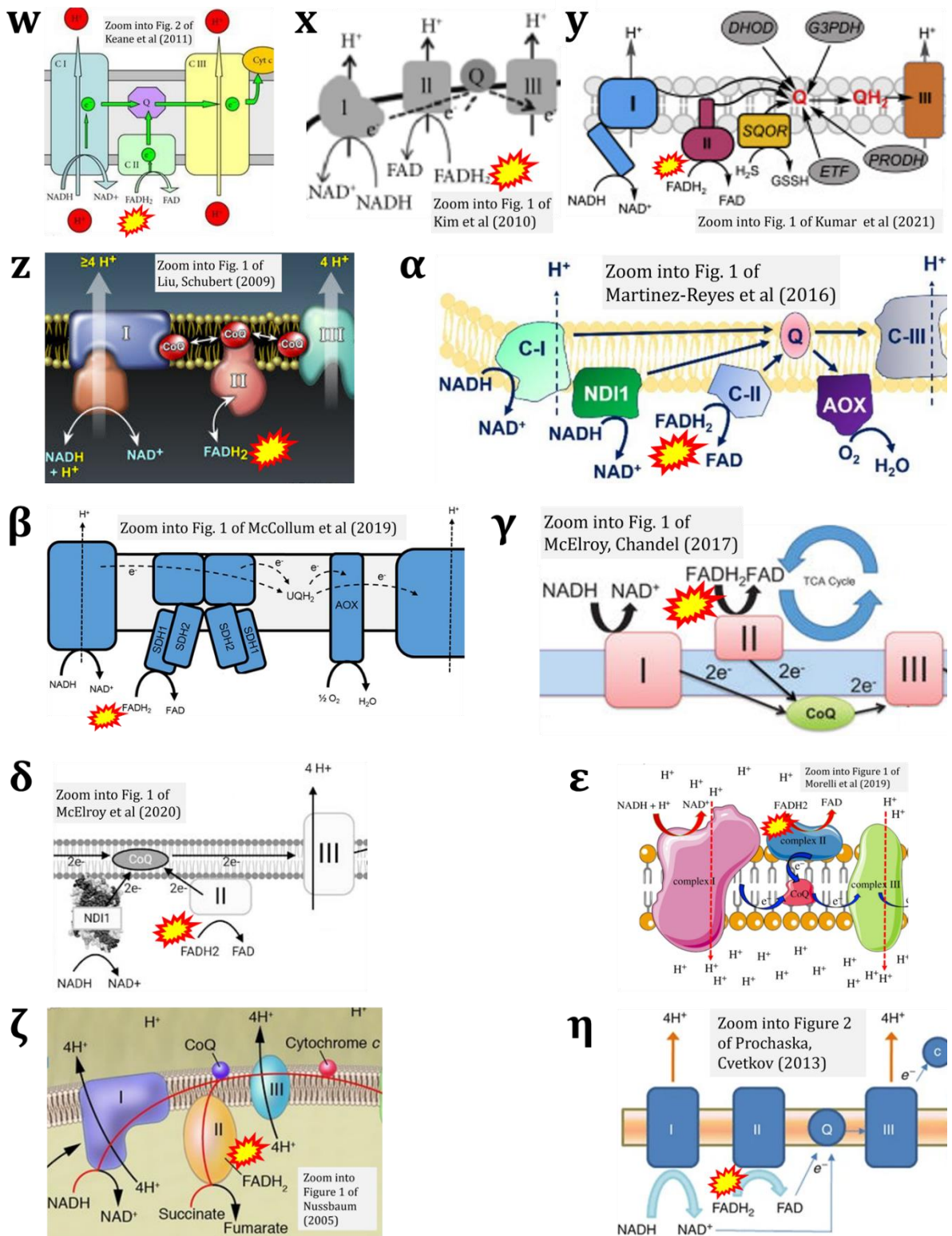


Figure S3. Continued

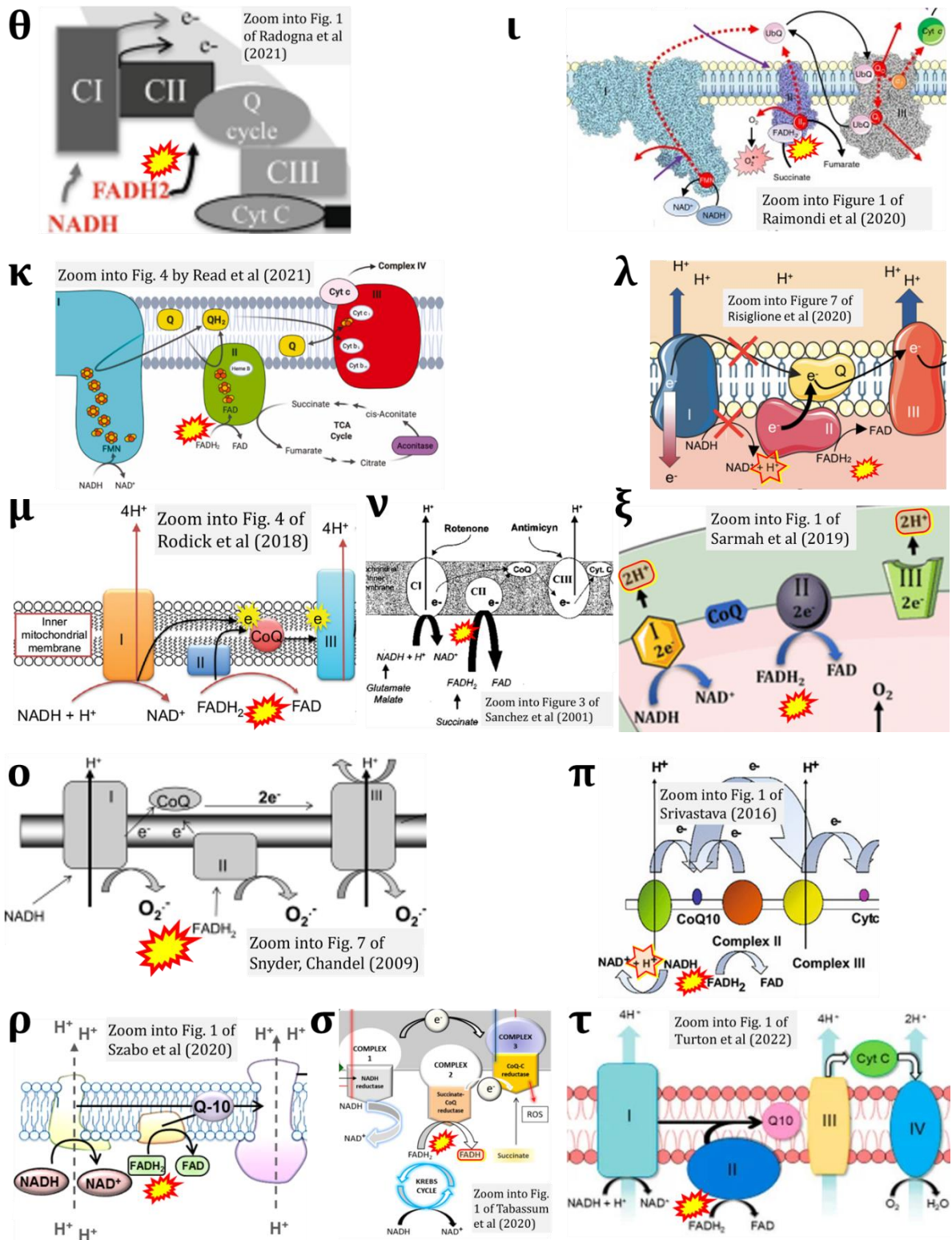


Figure S3. Continued

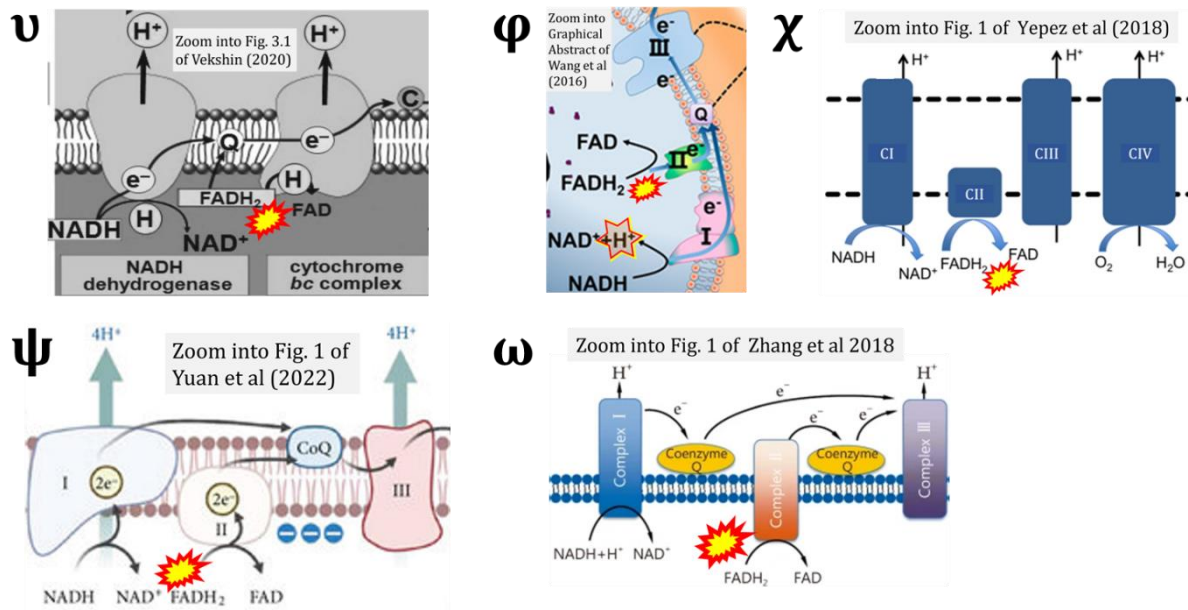


Figure S3. Complex II ambiguities in graphical representations on FADH₂ as a substrate of Complex II in the canonical forward electron transfer. Alphabetical sequence of publications from 2001 to 2023. See References for Figure S3.

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Supplement 4

FADH₂ as substrate of CII and FAD + 2H⁺ as products (Figure S4)

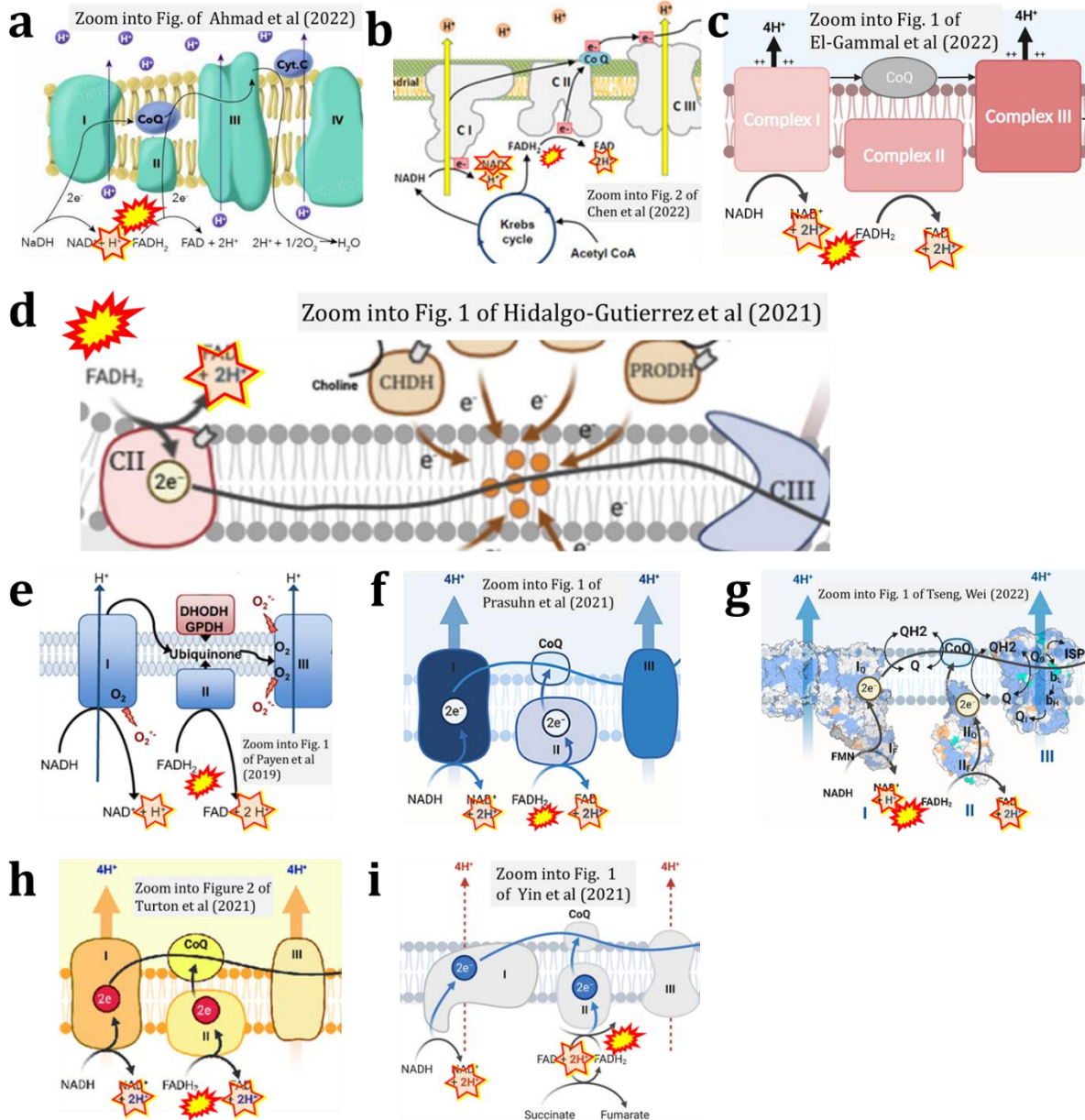
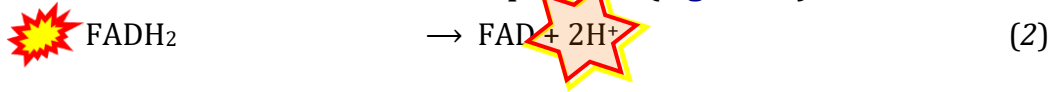


Figure S4. Complex II ambiguities: FADH₂ as substrate of CII and FAD + 2H⁺ as products. Alphabetical sequence of publications from 2001 to 2023. See References for Figure S4.

References for Figure S4

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Supplement 5

FADH₂ as substrate of CII and FAD⁺ as product (Figure S5)

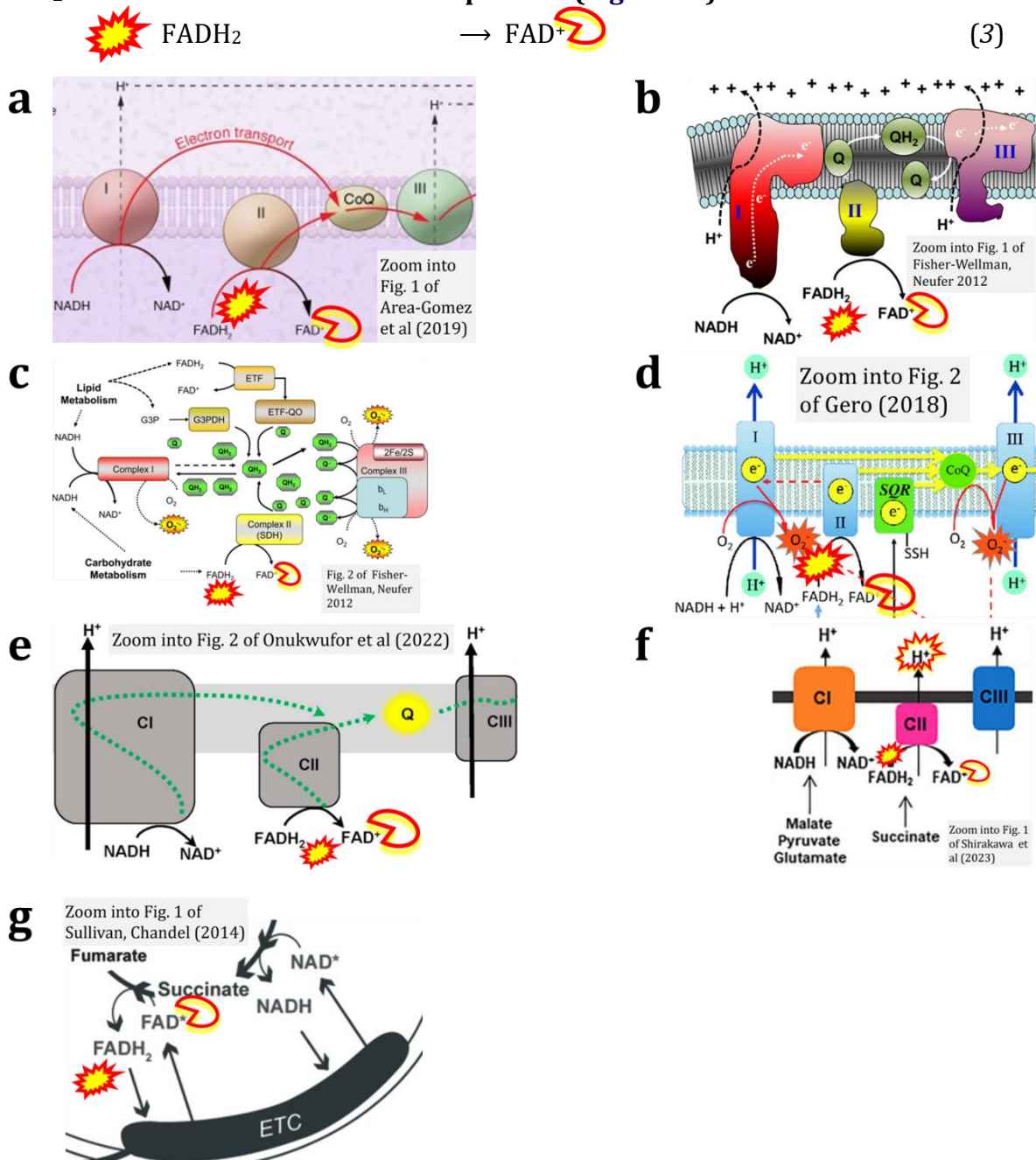


Figure S5. Complex II ambiguities: FADH₂ or FADH as substrate of CII and FAD⁺ or FAD⁺ + H⁺ as products. Alphabetical sequence of publications from 2001 to 2023. See References for Figure S5.








References for Figure S5

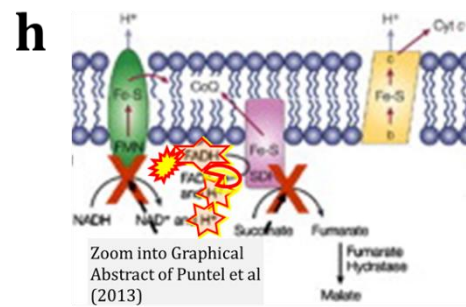
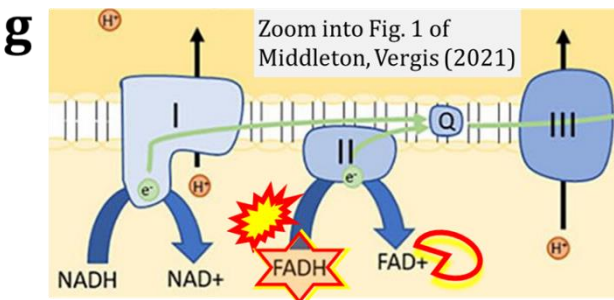
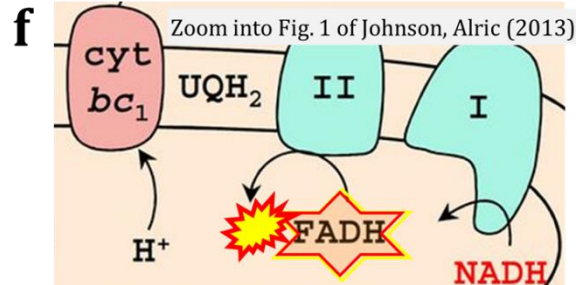
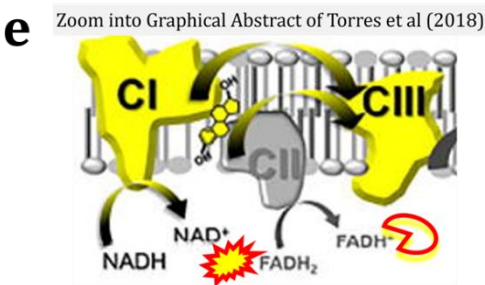
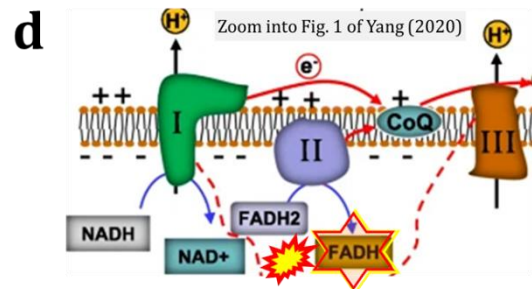
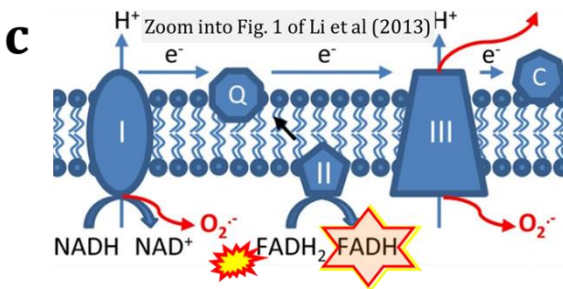
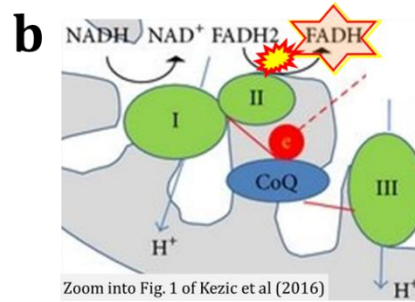
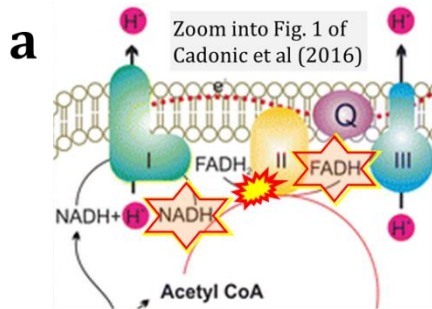
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Supplement 6

FADH₂ or FADH as substrate of CII and FADH, FADH⁺, or FAD⁺ as product (Figure S6)

- a-d  FADH₂ →  FADH (4)
- e  FADH₂ → FADH⁺ (5)
- f  FADH → (6)
- g  FADH → FAD⁺ (7)
- h  FADH → FAD⁺ + H⁺ (8)
- i  FADH → FAD⁺ + 2H⁺ (9)



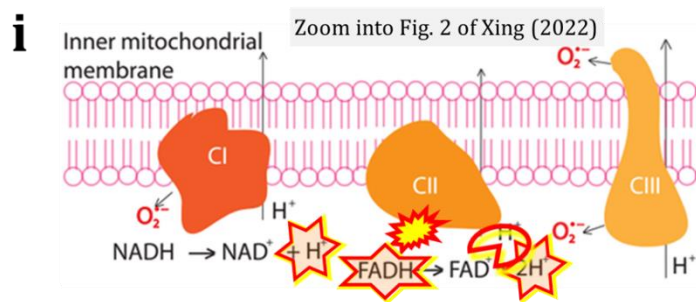


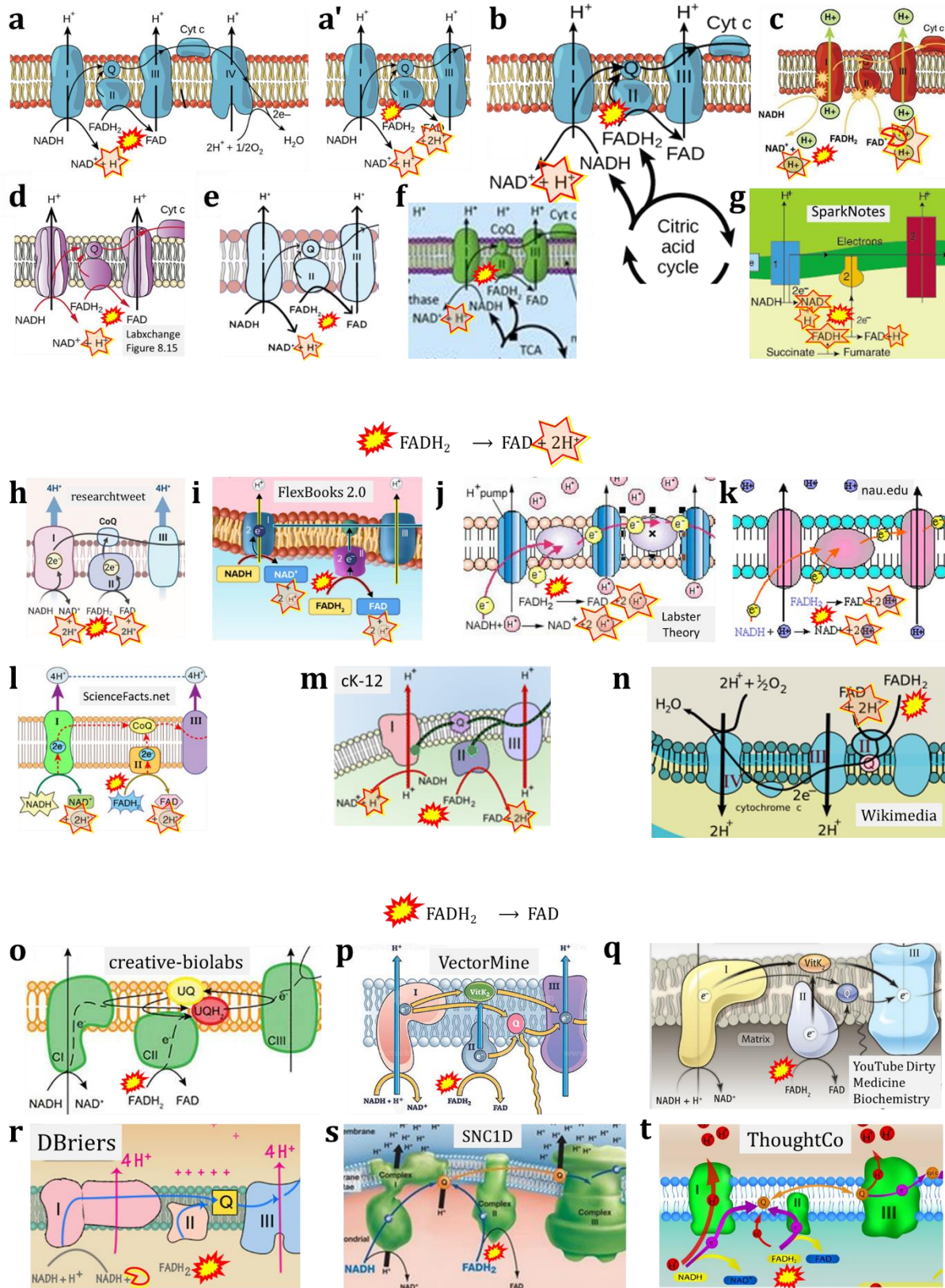
Figure S6. Complex II ambiguities: FADH₂ as substrate of CII and FADH or FADH⁺ as product. Sequence of publications from 2001 to 2023 according to FAD-*a* to FAD-*h*. See References for Figure S6.

References for Figure S6

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Supplement 7

FADH₂ or FADH as substrate of CII in websites (Figure S7)



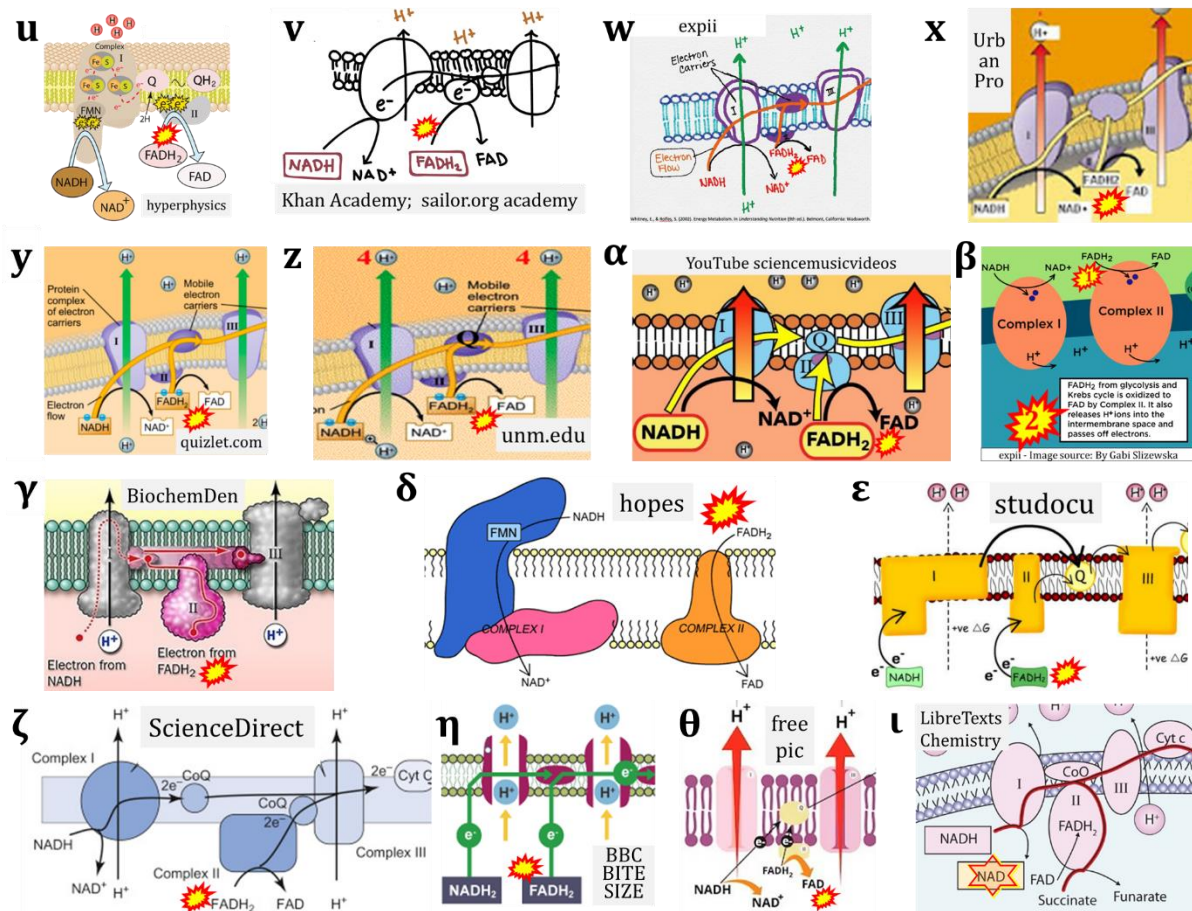


Figure S7. Complex II ambiguities in graphical representations on FADH₂ as a substrate of Complex II in the canonical forward electron transfer. FADH → FAD+H (g), FADH₂ → FAD+2H⁺ (a', c, h-n), and FADH₂ → FAD (a, b, d-f, o-θ) should be corrected to FADH₂ → FAD (Eq. 3b). NADH → NAD⁺ is frequently written in graphs without showing the H⁺ on the left side of the arrow, except for (p-r). NADH → NAD⁺+H⁺ (a-g, m), NADH → NAD⁺+2H⁺ (h-l), NADH+H⁺ → NAD⁺+2H⁺ (j, k), and NADH → NAD (t) should be corrected to NADH+H⁺ → NAD⁺ (Eq. 3a). Weblinks #: (a) 1-5, 8-10; (a') 6-7; (b) 6-9,11; (c) 10-16,35; (d) 17; (e) 18; (f) 19; (g) 20; (h) 21-22; (i) 23; (j) 24; (k) 25; (l) 26; (m) 27; (n) 11,28; (o) 29; (p) 30-31; (q) 32; (r) 33; (s) 34; (t) 22,35; (u) 36; (v) 6-7; (w) 11; (x) 37; (y) 38; (z) 39; (α) 40; (β) 11; (γ) 41; (δ) 42; (ε) 43; (ζ) 44; (η) 45; (θ) 46; (i) 47.

Weblinks for Figure S7 (retrieved 2023-03-21 to 2023-05-04)

- 1 (a,b) <https://openstax.org/books/biology/pages/7-4-oxidative-phosphorylation> - OpenStax Biology (CC BY 3.0) - Fig. 7.10 / Fig. 7.12
- 2 (a) <https://opentextbc.ca/biology/chapter/4-3-citric-acid-cycle-and-oxidative-phosphorylation/> - Charles Molnar, Jane Gair, Concepts of Biology - 1st Canadian Edition, BCCampus - Fig. 4.19a
- 3 (a) [https://bio.libretexts.org/Bookshelves/Introductory_and_General_Biology/Book%3AGeneral_Biology_\(Boundless\)/07%3ACellular_Respiration/7.11%3AOxidative_Phosphorylation_-_Electron_Transport_Chain](https://bio.libretexts.org/Bookshelves/Introductory_and_General_Biology/Book%3AGeneral_Biology_(Boundless)/07%3ACellular_Respiration/7.11%3AOxidative_Phosphorylation_-_Electron_Transport_Chain) - LibreTexts Biology - Fig. 7.11.1
- 4 (a) <https://www.pharmaguideline.com/2022/01/electron-transport-chain.html> - Pharmaguideline
- 5 (a) <https://brainbrooder.com/lesson/254/7-4-1-electron-transport-chain> - Brain Brooder

- 6 (a',b,v) <https://www.khanacademy.org/science/ap-biology/cellular-energetics/cellular-respiration-ap/a/oxidative-phosphorylation-etc> - Khan Academy - Image modified from "Oxidative phosphorylation: Fig. 1", by OpenStax College, Biology (CC BY 3.0) / Image modified from "Oxidative phosphorylation: Fig. 3," by Openstax College, Biology (CC BY 3.0)
- 7 (a',b,v) <https://learn.saylor.org/mod/page/view.php?id=32815> -Saylor Academy
- 8 (a,b) <https://www.texasgateway.org/resource/74-oxidative-phosphorylation> - Texas Gateway - Figure 7.11
- 9 (a,b) <https://opened.cuny.edu/courseware/lesson/639/overview> -CUNY
- 10 (a,c) <https://courses.lumenlearning.com/wm-biology1/chapter/reading-electron-transport-chain/> - lumen Biology for Majors I - Fig. 1 / Fig. 3
- 11 (b,c,w,n,β) <https://www.expil.com/t/electron-transport-chain-summary-diagrams-10139> - expil - Image source: By CNX OpenStax / By OpenStax College CC BY 3.0, via Wikimedia Commons / Whitney, Rolfes 2002 / By User:Rozzychan CC BY-SA 2.5, via Wikimedia Commons
- 12 (c) <https://commons.wikimedia.org/w/index.php?curid=30148497> - wikimedia 30148497 - Anatomy & Physiology, Connexions Web site. <http://cnx.org/content/col11496/1.6/>, 2013-06-19
- 13 (c) <https://biologydictionary.net/electron-transport-chain-and-oxidative-phosphorylation/> - biologydictionary.net 2018-08-21
- 14 (c) <https://www.quora.com/Why-does-FADH2-form-2-ATP> - Quora
- 15 (c) <https://teachmephysiology.com/biochemistry/atp-production/electron-transport-chain/> - TeachMePhysiology - Fig. 1. 2023-03-13
- 16 (c) <https://www.toppr.com/ask/question/short-long-answer-types-what-is-the-electron-transport-system-and-what-are-its-functions/> - toppr
- 17 (d) <https://www.labxchange.org/library/items/lb:LabXchange:005ad47f-7556-3887-b4a6-66e74198fbcf:html:1> - Labxchange - Figure 8.15 credit: modification of work by Klaus Hoffmeier
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- 19 (f) <https://videodelivery.net/79e91c40bf96f9692560fa378c5086b6/thumbnails/thumbnail.jpg> - videodelivery
- 20 (g) <https://www.sparknotes.com/biology/cellrespiration/oxidativephosphorylation/section2/> - SparkNotes
- 21 (h) <https://microbenotes.com/electron-transport-chain/> - Microbe Notes
- 22 (h,t) <https://researchtweet.com/mitochondrial-electron-transport-chain-2/> - researchtweet
- 23 (i) <https://flexbooks.ck12.org/cbook/ck-12-biology-flexbook-2.0/section/2.28/primary/lesson/electron-transport-bio/> - FlexBooks - CK-12 Biology for High School- 2.28 Electron Transport, Fig. 2
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- 25 (k) <https://www2.nau.edu/~fpm/bio205/u4fg36.html> - nau.edu
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- 28 (n) https://commons.wikimedia.org/wiki/File:Mitochondrial_electron_transport_chain.png - Wikimedia
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- 30 (p) <https://www.dreamstime.com/electron-transport-chain-as-respiratory-embedded-transporters-outline-diagram-electron-transport-chain-as-respiratory-embedded-image235345232> - dreamstime
- 31 (p) <https://vectormine.com/item/electron-transport-chain-as-respiratory-embedded-transporters-outline-diagram/> - VectorMine
- 32 (q) https://www.google.com/imgres?imgurl=https%3A%2F%2Fi.ytimg.com%2Fvi%2FLsRQ5_EmxJA%2Fmaxresdefault.jpg&tbnid=6w-0DVPMw7vOdM&vet=12ahUKEwjw2Y05--T9AhUwpCcCHduuDVGQMygDegUIARDzAQ..i&imgrefurl=https%3A%2F%2Fwww.youtube.com%2Fwatch%3Fv%3DLsRQ5_EmxJA&docid=bZxQYNch1Ys-VM&w=1280&h=720&q=electron%20transport%20chain&hl=en-US&client=firefox-b-d&ved=2ahUKEwjw2Y05--T9AhUwpCcCHduuDVGQMygDegUIARDzAQ - YouTube Dirty Medicine Biochemistry - Uploaded 2019-07-18
- 33 (r) <http://www.dbriers.com/tutorials/2012/04/the-electron-transport-chain-simplified/> - DBriers
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- 36 (u) <http://hyperphysics.phy-astr.gsu.edu/hbase/Biology/Complex1.html> - hyperphysics
- 37 (x) <https://www.urbanpro.com/ba-tuition/oxidative-phosphorylation> - UrbanPro
- 38 (y) <https://quizlet.com/245664214/electron-transport-chain-facts-of-cell-respiration-diagram/> - Quizlet
- 39 (z) <https://www.unm.edu/~lkravitz/Exercise%20Phys/ETCstory.html> - unm.edu
- 40 (α) https://www.google.com/imgres?imgurl=https%3A%2F%2Fi.ytimg.com%2Fvi%2FVER6xW_r1vc%2Fmaxresdefault.jpg&tbnid=Brshl0oN9LyYnM&vet=12ahUKEwjKSKpOX9AhWjmycCHbvGC34QMygWegUIARDWAQ..i&imgrefurl=https%3A%2F%2Fwww.youtube.com%2Fwatch%3Fv%3DVER6xW_r1vc&docid=VgTgrLf24Lzg4M&w=1280&h=720&itg=1&q=FADH2%20is%20the%20substrates%20of%20Complex%20II&hl=en&client=firefox-b-d&ved=2ahUKEwjKSKpOX9AhWjmycCHbvGC34QMygWegUIARDWAQ - YouTube sciencemusicvideos - Uploaded 2014-08-19
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- 42 (δ) <https://hopes.stanford.edu/riboflavin/> - hopes, Huntington's outreach project for education, at Stanford
- 43 (ε) <https://www.studocu.com/en-gb/document/university-college-london/mammalian-physiology/electron-transport-chain/38063777> - studocu, University College London
- 44 (ζ) https://www.google.com/imgres?imgurl=https%3A%2F%2Fars.els-cdn.com%2Fcontent%2Fimage%2F3-s2.0-B9780128008836000215-f21-07-9780128008836.jpg&imgrefurl=https%3A%2F%2Fwww.sciencedirect.com%2Ftopics%2Fengineering%2Felectron-transport-chain&tbnid=g3dD4u8Tvd6TWM&vet=12ahUKEwjc9deUprT9AhVxhv0HHXZbAd0QMygCegUIARDBAQ..i&docid=Moj_2_W0OpUDcM&w=632&h=439&q=FADH2%20is%20the%20substrates%20of%20Complex%20II&client=firefox-b-d&ved=2ahUKEwjc9deUprT9AhVxhv0HHXZbAd0QMygCegUIARDBAQ - ScienceDirect
- 45 (η) <https://www.bbc.co.uk/bitesize/guides/zdq9382/revision/5> - BBC BITESIZE
- 46 (θ) https://www.freepik.com/premium-vector/oxidative-phosphorylation-process-electron-transport-chain-final-step-cellular-respiration_29211885.htm - freepik
- 47 (ι) [https://chem.libretexts.org/Courses/Saint_Marys_College_Notre_Dame_IN/CHEM_118_\(Under_Construction\)/CHEM_118_Textbook/12%3AMetabolism_\(Biological_Energy\)/12.4%3A_The_Citric_Acid_Cycle_and_Electron_Transport](https://chem.libretexts.org/Courses/Saint_Marys_College_Notre_Dame_IN/CHEM_118_(Under_Construction)/CHEM_118_Textbook/12%3AMetabolism_(Biological_Energy)/12.4%3A_The_Citric_Acid_Cycle_and_Electron_Transport) - LibreTexts Chemistry - The Citric Acid Cycle and Electron Transport – Fig. 12.4.3

Supplement 8

Weblinks on FAO and CII (retrieved 2023-03-21 to 2023-05-02)

- 48 <https://conductscience.com/electron-transport-chain/> - Conduct Science: "*In Complex II, the enzyme succinate dehydrogenase in the inner mitochondrial membrane reduce FADH₂ to FAD⁺. Simultaneously, succinate, an intermediate in the Krebs cycle, is oxidized to fumarate.*" - Comments: FAD does not have a positive charge. FADH₂ is the reduced form, it is not reduced. And again: *In CII, FAD is reduced to FADH₂.*
- 49 <https://themedicalbiochemistrypage.org/oxidative-phosphorylation-related-mitochondrial-functions/> - The Medical Biochemistry Page: '*In addition to transferring electrons from the FADH₂ generated by SDH, complex II also accepts electrons from the FADH₂ generated during fatty acid oxidation via the fatty acyl-CoA dehydrogenases and from mitochondrial glycerol-3-phosphate dehydrogenase (GPD2) of the glycerol phosphate shuttle*' (Figure 8d).
- 50 <https://www.chem.purdue.edu/courses/chm333/Spring%202013/Lectures/Spring%202013%20Lecture%2037%20-%2038.pdf> - CHM333 LECTURES 37 & 38: 4/27 – 29/13 SPRING 2013 Professor Christine Hrycyna - Acyl-CoA dehydrogenase is listed under 'Electron transfer in Complex II'.

Supplement 9

CII as a proton pump (Figure S9)

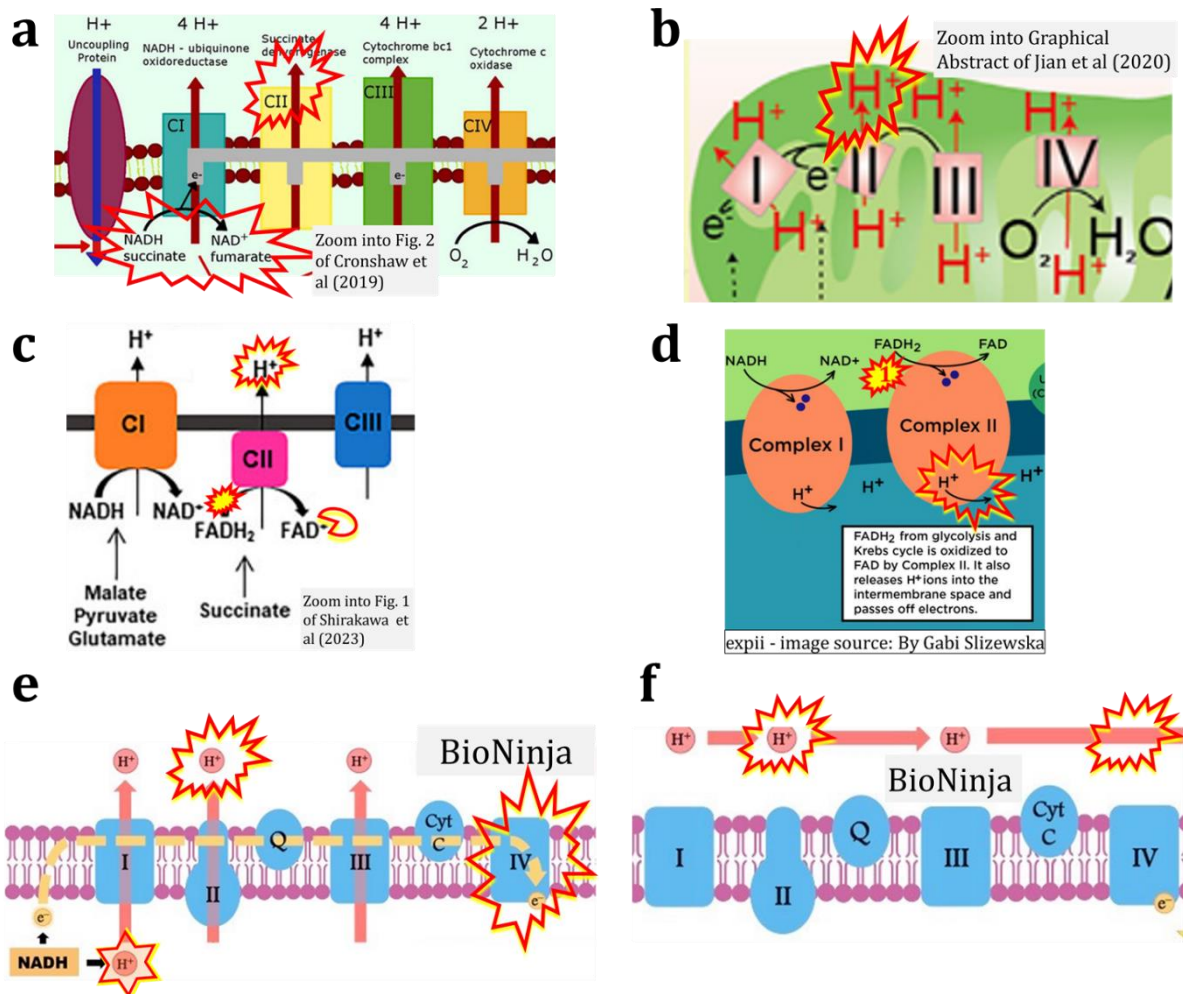


Figure S9. Complex II as a proton pump.

- a** Cronshaw M, Parker S, Arany P (2019) Feeling the heat: evolutionary and microbial basis for the analgesic mechanisms of photobiomodulation therapy. **Photobiomodul Photomed Laser Surg** 37:517-26. <https://doi.org/10.1089/photob.2019.4684>
- b** Jian C, Fu J, Cheng X, Shen LJ, Ji YX, Wang X, Pan S, Tian H, Tian S, Liao R, Song K, Wang HP, Zhang X, Wang Y, Huang Z, She ZG, Zhang XJ, Zhu L, Li H (2020) Low-dose sorafenib acts as a mitochondrial uncoupler and ameliorates nonalcoholic steatohepatitis. **Cell Metab** 31:892-908. <https://doi.org/10.1016/j.cmet.2020.04.011>
- c** Shirakawa R, Nakajima T, Yoshimura A, Kawahara Y, Orito C, Yamane M, Handa H, Takada S, Furihata T, Fukushima A, Ishimori N, Nakagawa M, Yokota I, Sabe H, Hashino S, Kinugawa S, Yokota T (2023) Enhanced mitochondrial oxidative metabolism in peripheral blood mononuclear cells is associated with fatty liver in obese young adults. **Sci Rep** 13:5203. <https://doi.org/10.1038/s41598-023-32549-w>
- d** <https://www.exprii.com/t/electron-transport-chain-summary-diagrams-10139> - exprii - Image source: By Gabi Slizewska: 'FADH₂ from glycolysis and Krebs cycle is oxidized to FAD by Complex II. It also releases H⁺ ions into the intermembrane space and passes off electrons' (retrieved 2023-05-04).
- e,f** <https://ib.bioninja.com.au/higher-level/topic-8-metabolism-cell/untitled/electron-transport-chain.html> - BioNinja (retrieved 2023-05-04).